

Methyl 2,6-bis(4,6-dimethoxypyrimidin-2-yloxy)benzoate

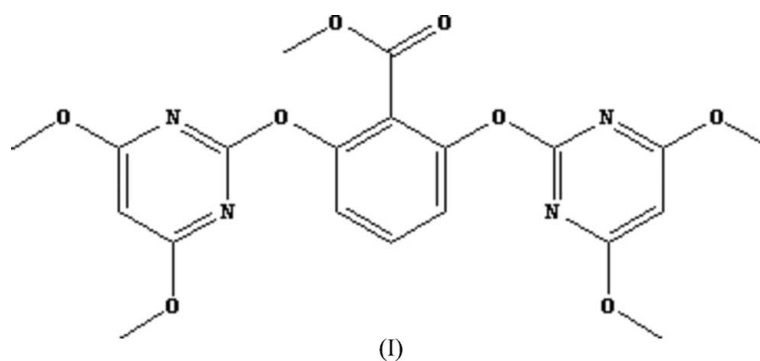
Yuan-Xiang Li* and Yan-Ping Luo

Key Laboratory of Pesticides and Chemical
Biology of the Ministry of Education, College of
Chemistry, Central China Normal University,
Wuhan 430079, People's Republic of ChinaCorrespondence e-mail:
liyuanjun68@yahoo.com.cn

Key indicators

Single-crystal X-ray study
 $T = 292$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.050
 wR factor = 0.129
Data-to-parameter ratio = 14.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.In the title compound, $\text{C}_{20}\text{H}_{20}\text{N}_4\text{O}_8$, there are intermolecular
 $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds and $\pi-\pi$ stacking interactions.Received 23 February 2006
Accepted 6 March 2006

Comment

KIH-2023, a phenoxypyrimidine derivative, is a highly
effective herbicide (Wada *et al.*, 1988, 1990, 1992). In this
paper, we report the result of an X-ray diffraction study of the
title compound, (I), which has a similar structure to the
herbicide KIH-2023.

The two planar pyrimidine rings are twisted (Fig. 1), the angle between them being $87.6(2)^\circ$. The dihedral angles between the benzene ring and the pyrimidine (N1/C3–C5/N2/C6) and (N3/C15/N4/C16–C18) rings are $65.7(2)$ and $58.1(2)^\circ$, respectively. The crystal packing is reinforced by an intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bond (Table 2 and Fig. 2) and a $\pi-\pi$ stacking interaction between the benzene rings [$\text{C}7-\text{C}12$ and $\text{C}7^{\text{ii}}-\text{C}12^{\text{ii}}$; symmetry code: (ii) $-x, 2-y, -z$] (Fig. 3). The centroid–centroid distance is $3.917(4)$ Å and the interplanar spacing is $3.398(4)$ Å.

Experimental

2,6-Dihydroxybenzoic acid (15.4 g, 0.1 mol) in methanol (60 ml) with H_2SO_4 (3.5 ml) was refluxed for 34 h. The excess methanol was removed under reduced pressure and 300 ml water was then added. The precipitate was filtered off and dried at room temperature. The product was recrystallized from cyclohexane to give methyl 2,6-dihydroxybenzoate (7.3 g, yield 43.5%). To a solution of this ester (3 g, 17.8 mmol) in acetone (50 ml) was added K_2CO_3 (6.16 g, 44.6 mmol) and the mixture was refluxed for 1 h. 2-Methanesulfonyl-4,6-dimethoxypyrimidine (7.79 g, 35.73 mmol) was added and the mixture was refluxed for 5 h. The precipitate was filtered off and dried under reduced pressure to give a white solid which was purified by recrystallization from ethanol (Yield 7.25 g, 91.5%). Crystals suitable for X-ray diffraction were obtained by slow evaporation of an ethanol solution at 298 K.

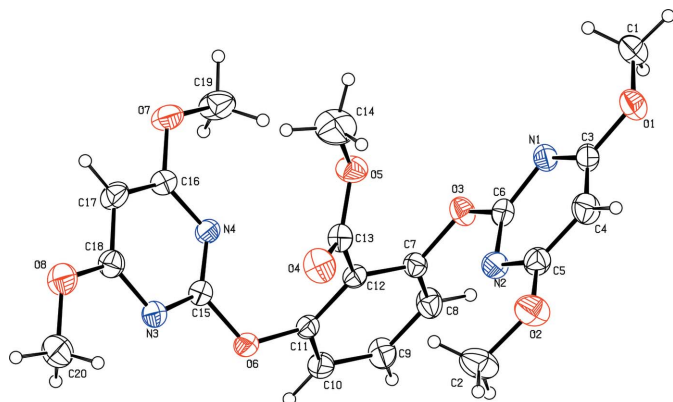


Figure 1
A view of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level.

Crystal data

$C_{20}H_{20}N_4O_8$	$D_x = 1.393 \text{ Mg m}^{-3}$
$M_r = 444.40$	Mo $K\alpha$ radiation
Monoclinic, $C2/c$	Cell parameters from 1421 reflections
$a = 29.00 (3) \text{ \AA}$	$\theta = 2.8\text{--}21.6^\circ$
$b = 7.776 (7) \text{ \AA}$	$\mu = 0.11 \text{ mm}^{-1}$
$c = 22.34 (2) \text{ \AA}$	$T = 292 (2) \text{ K}$
$\beta = 122.753 (16)^\circ$	Block, colorless
$V = 4237 (7) \text{ \AA}^3$	$0.30 \times 0.20 \times 0.10 \text{ mm}$
$Z = 8$	

Data collection

Bruker SMART 4K CCD area-detector diffractometer	2544 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\text{int}} = 0.046$
Absorption correction: none	$\theta_{\text{max}} = 26.0^\circ$
11217 measured reflections	$h = -21 \rightarrow 35$
4161 independent reflections	$k = -9 \rightarrow 9$
	$l = -27 \rightarrow 25$

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.050$	$w = 1/[\sigma^2(F_o^2) + (0.0542P)^2]$
$wR(F^2) = 0.129$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 0.99$	$(\Delta/\sigma)_{\text{max}} < 0.001$
4161 reflections	$\Delta\rho_{\text{max}} = 0.15 \text{ e \AA}^{-3}$
294 parameters	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

C1—O1	1.440 (3)	C6—O3	1.364 (3)
C3—O1	1.344 (3)	C13—O4	1.194 (3)
C6—N1	1.325 (3)	C13—O5	1.337 (3)
N2—C6—O3	118.5 (2)	C12—C7—O3	122.2 (2)
N1—C6—O3	111.9 (2)	O5—C13—C12	111.4 (2)
C8—C7—O3	115.8 (2)		

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
C14—H14C \cdots O7 ⁱ	0.96	2.60	3.444 (4)	147

Symmetry code: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$.

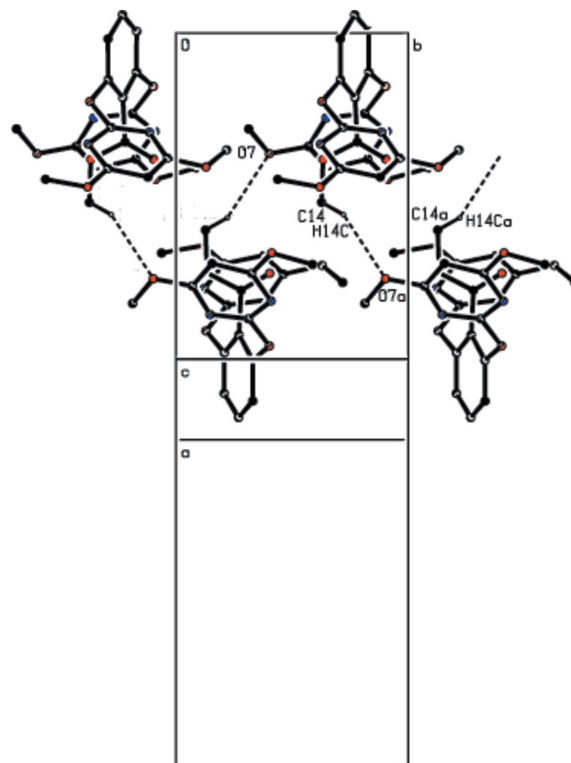


Figure 2
C—H \cdots O hydrogen bonds in (I). Hydrogen bonds are shown as dashed lines. The suffix a corresponds to symmetry code (i) in Table 2.

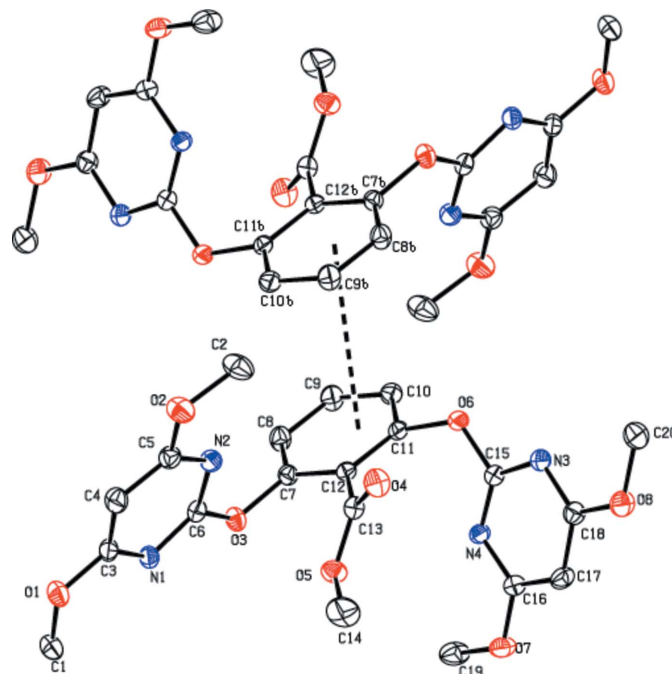


Figure 3
A $\pi\text{--}\pi$ stacking interaction (dashed line) in (I). The suffix b corresponds to symmetry code $(-x, 2 - y, -z)$.

H atoms were placed at calculated positions and refined as riding (C—H = 0.93–0.96 \AA), with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1999); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.

The authors acknowledge financial support from the National Key Project for Basic Research (No. 2002CCA00500), the National Natural Science Foundation of China (Nos. 20432010, 20476036 and 20172017), the Program for New Century Excellent Talents in Universities of China

and the Program for Excellent Research Groups of Hubei Province (No. 2004ABC002).

References

- Bruker (1997). *SMART*. Version 5.054. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (1999). *SAINTE*. Version 6.01. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2001). *SHELXTL*. Version 6.12. Bruker AXS Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Wada, N., Kusano, S. & Toyokawa, Y. (1988). EP Patent No. 321846.
- Wada, N., Kusano, S. & Toyokawa, Y. (1990). US Patent No. 4906285.
- Wada, N., Kusano, S. & Toyokawa, Y. (1992). US Patent No. 5081244.